

DEMYSTIFYING THE SOAI REACTION

Soumitra V. Athavale^a, Adam Simon^b, Kendall N. Houk^b and Scott E. Denmark^a

^aRoger Adams Laboratory, Department of Chemistry, University of Illinois, Urbana, Illinois, 61801, United States

^bDepartment of Chemistry and Biochemistry, University of California, Los Angeles, California, 90095, United States

The extraordinary Soai reaction has profoundly impacted chemists' perspective of chiral symmetry breaking, absolute asymmetric synthesis and its role in the origin of biological homochirality [1]. Herein, we describe the unprecedented observation of asymmetry amplifying autocatalysis in the alkylation of 5-(trimethylsilylethynyl)pyridine-3-carbaldehyde using diisopropylzinc (Figure 1a). Kinetic studies with a "Trojan-horse" substrate and spectroscopic analysis of a series of zinc-alkoxides that incorporate specific structural mutations reveal a 'pyridine-assisted cube escape' (Figure 1c). The new cluster functions as a catalyst that activates the 'floor-to-floor' bound aldehyde and poises a coordinated diisopropylzinc moiety for alkyl group transfer. Transition-state models leading to both the homochiral and heterochiral products were validated by density functional theory calculations (Figure 1b). Moreover, experimental and computational analysis of the heterochiral complex provides a definitive explanation for the non-linear behavior of this system. Our deconstruction of the Soai system contributes substantially to understanding the mechanism of this transformation that has stood as a longstanding challenge in chemistry.

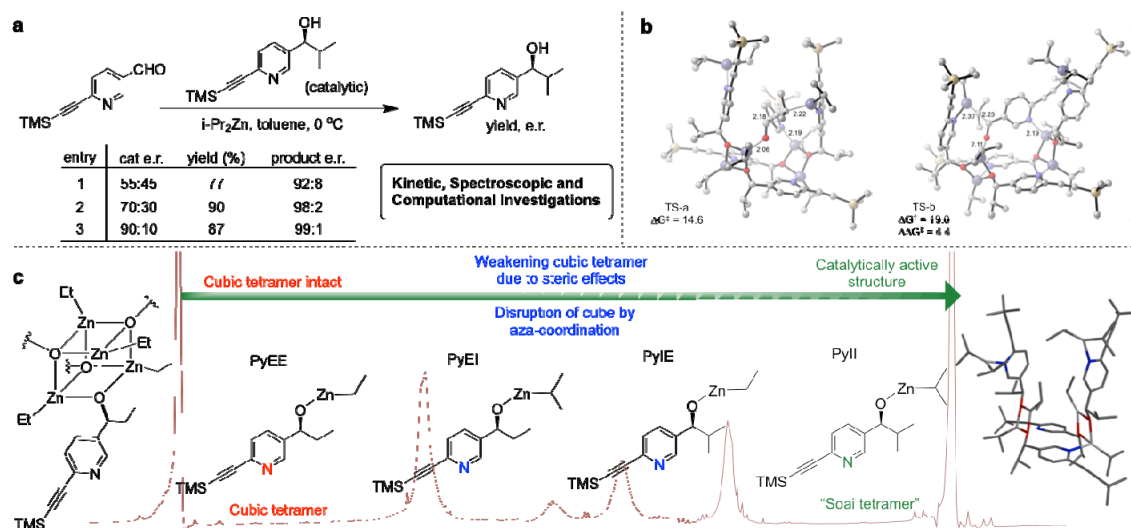


Figure 1: a. Amplifying autocatalysis with 5-(trimethylsilylethynyl)pyridine-3-carbaldehyde and diisopropylzinc.

b. Transition state models leading to the homochiral (TS-a) and heterochiral product (TS-b) c. Evolution of the Soai tetramer due to pyridine coordination assisted 'cube escape'.