## COPPER-CATALYZED BORYLATION OF OLEFINS: A QUEST FOR REACTIVITY AND SELECTIVITY

Daniele Fiorito, Yangbin Liu and Clément Mazet\*

## Department of Organic Chemistry, University of Geneva Quai Ernest Ansermet 30, 1211 Geneva 4, Switzerland clement.mazet@unige.ch

Copper-catalyzed borylation of olefins represents a cutting-edge methodology for the introduction of boron in organic molecules. Migratory insertion of alkenes across a copper-boron bond generates nucleophilic alkyl copper intermediates, which display broad reactivity towards electrophiles.<sup>[1]</sup> Proton is the simplest electrophile that can be employed and the resulting protoboration of olefins complements hydroboration approaches.<sup>[2]</sup> Furthermore, the use of carbon based electrophiles offers new strategies for the construction of C–C bonds.<sup>[3]</sup> Herein we describe our results in the identification of two novel reactions in the context of copper-borylation of olefins. As a continuation of our efforts in the carbofunctionalization of alkenes,<sup>[4]</sup> we developed a Cu-catalyzed borylative carboxamidation of olefins using isocyanates as electrophiles. The method allows expedient access to chiral secondary amides bearing a boronate handle starting from commodity chemicals and a readily accessible [(NHC)Cu] precatalyst. Preliminary results for the enantioselective version of the reaction will be disclosed.<sup>[5]</sup> Amongst conjugated alkenes, 1,3-dienes represent an underexplored class of substrates for selective transition metal catalysis. This is due to the numerous coordination and insertion modes conceivable for a transient organometallic species. In this context, we devised a Cu-catalyzed 1,2-protoboration of 2-substituted 1,3-dienes with exquisite anti-Markovnikov selectivity. The use of a chiral phosphanamine ligand is essential in achieving high chemo-, regio- and enantioselectivity, providing rapid access to a variety of synthetically relevant homoallylic boronates.<sup>[6]</sup>



<sup>[1]</sup> E. C. Neeve, S. J. Geier, I. A. I. Mkhalid, S. A. Westcott and T. B. Marder, *Chem. Rev.*, **2016**, *116*, 9091.

<sup>[2] (</sup>a) J.-E. Lee and J. Yun, Angew. Chem., Int. Ed., 2008, 47, 145; (b) Y. Sasaki, C. Zhong, M. Sawamura and H. Ito, J. Am. Chem. Soc., 2010, 132, 1226.

<sup>[3] (</sup>a) T. W. Butcher, E. J. McClain, T. G. Hamilton, T. M. Perrone, K. M. Kroner, G. C. Donohoe, N. G. Akhmedov, J. L. Petersen and B. V. Popp, *Org. Lett.*, **2016**, *18*, 6428; (b) Y. Huang, K. B. Smith and M. Kevin Brown, *Angew. Chem. Int. Ed.*, **2017**, *56*, 13314.

<sup>[4]</sup> G. M. Borrajo-Calleja, V. Bizet and C. Mazet, J. Am. Chem. Soc., 2016, 138, 4014

<sup>[5]</sup> D. Fiorito, Y. Liu and C. Mazet, manuscript in preparation.

<sup>[6]</sup> Y. Liu, D. Fiorito and C. Mazet, Chem. Sci., 2018, 9, 5284.