## ASYMMETRIC LACTONE SYNTHESIS BY REDUCTIVE OLEFIN ACTIVATION AND CO<sub>2</sub> -FIXATION

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Olefins are an attractive coupling partner with carbon dioxide due their wide availability, low cost and high atom effiency. In previous studies, olefins have been activated by reaction with a strong nucleophile (hydride[1], organometal reagents[2]) to form strongly nucleophilic carbon, that can further react with carbon dioxide. In our research[3], we have found that metals with high reduction potential can force the olefin to react twice as a nucleophile. This reactivity was used to synthesize 3,4-disubstituted five-membered lactones in moderate to high yields and high diastereoselectivity. We anticipate that this type of reactivity could be useful in a variety of synthesis routes, allowing fast increase of complexity, with formation of two C-C bonds and two chiral centers.

$$R_1$$
 +  $R_2$  H +  $C_1$  +  $C_2$   $M(Ligand^*)$   
Reductant  $R_2$   $R_1$ 

<sup>[1]</sup> Catherine M. Williams, Jeffrey B. Johnson, and Tomislav Rovis, J. Am. Chem. Soc. 2008 130 (45), 14936-14937; Shingo Kawashima, Kohsuke Aikawa, and Koichi Mikami, Eur. J. Org. Chem. 2016, 3166–3170

<sup>[2]</sup> Anne-Marie L. Hogan and Donal F. O'Shea, J. Org. Chem. 2007, 72, 9557-9571; Peng Shao, Sheng Wang, Chao Chen and Chanjuan Xi, Chem. Commun., 2015, 51, 6640--6642

<sup>[3]</sup> Aleksi Sahari, Jere K. Mannisto and Timo Repo, Asymmetric lactone synthesis by reductive olefin activation and  $CO_2$  –fixation, Manuscript in preparation